WATER QUALITY SURVEY

OF THE

PATUXENT RIVER

1969

DATA REPORT

Number 17

Annapolis Field Office Region III Environmental Protection Agency

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Staff

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I INTRODUCTION

A. Purpose and Scope

During the year of 1969 the Annapolis Field Office, Region III, Environmental Protection Agency continued surveys of the estuary portion of the Patuxent River. The purpose of the surveys was to continue upgrading data from previous years and model verification.

B. General Remarks

The Patuxent estuary was sampled six times during the year.

Monthly sample runs were made during the summer months of June through August.

C. Sampling Procedures

Samples for chemical analysis were obtained by dipping a plastic bucket into the river water with a minimum of agitation. The water sample was then transferred to a plastic one-quart cubitainer. Dissolved oxygen (DO) samples were obtained by siphoning from the bucket through plastic tubing extended to the bottom of a conventional 300 ml DO bottle. The bottle was overfilled twice without excessive agitation, and the DO fixed immediately. All samples were stored on ice. Analysis was started immediately on return to the Annapolis Field Office, generally within two hours of sampling.

D. Measured Parameters and Analytical Methods

- 1. Water temperature was determined using a mercury thermometer inserted into the sample container immediately upon collection.
 - 2. Water temperature was read from a Beckman Salinometer.
 - 3. Conductivity was read from a calibrated Beckman Salinometer.
- 4. Salinity was determined with a calibrated Beckman Salinometer.

5. Total Phosphorus

Reference: Menzel, D.W. and Corwin, N., 1965. The Measurement of Total Phosphorus in Seawater Based on the Liberation of Organically Bound Fractions by Persulfate Oxidation. Limonology and Oceanography, 10: 280-282.

Murphy, J. and Riley, J.P., 1962. A Modified Single Solution Method for the Determination of Phosphate in Natural Waters. Analytica Chimica Acta, 27: 31-36.

Total Phosphorus was determined after persulfate oxidation of the sample in an autoclave at 15 psi for 30 minutes. The resultant ortho-phosphate was then determined colorimetrically as the molybdenum-blue complex with the optical density measured at 882 mm.

6. Inorganic Phosphorus

Reference: Murphy, J. and Riley, J.P., 1962. A Modified

Single Solution Method for the Determination of Phosphate in Natural Waters. Analytica Chimica Acta, 27: 31-36.

Inorganic Phosphorus was determined by automation of the above procedure using the Technicon "AutoAnalyzer." The molybdenum-blue complex formed was determined colorimetrically with the optical density measured at $885~\text{m}\mu$.

7. Total Kjeldahl Nitrogen

Reference: Standard Methods for the Examination of Water and Wastewater, 12 ed., 1965.

Total Kjeldahl Nitrogen includes ammonia and organic nitrogen and was determined by the standard micro-kjeldahl procedure. The sample was digested in the presence of strong acid to convert the organic nitrogen to ammonia. The ammonia was then distilled, collected in boric acid solution, nesslerized and determined colorimetrically.

8. Nitrate and Nitrite

Reference: A Practical Handbook of Seawater Analysis, J.D.H. Strickland and T.R. Parsons, Bulletin 167, Fisheries Reasearch Board of Canada, Ottawa, Canada. 1968.

Nitrate plus nitrite nitrogen was determined by automation of the above procedure using the Technicon "Auto-Analyzer." This procedure utilizes cadmium reduction of nitrate to nitrite and subsequent diazotization with sulfanilamide

and N-(1-naphthyl)-ethylenediamine dihydrochloride with the optical density measured at 540 m μ . The results were reported as nitrogen.

9. Ammonia

Reference: Southeast Water Laboratory, FWQA,

Methodology for the colorimetric determination of ammonia
by the phenol-hypochlorite reaction.

FWPCA Methods for Chemical Analysis of Water and Wastes, November 1969.

Ammonia nitrogen was determined by automation of the phenol-hypochlorite procedure as described in the Southeast Water Laboratory Methodology and later adopted as the official FWPCA procedure. The intensity of the indophenol blue color, formed by the reaction of ammonia with alkaline phenol-hypochlorite, was increased using sodium nitroprusside as an intensifying agent. The optical density was measured at 630 mµ and calculated as NH₃-N.

10. Chlorophyll a

Reference: A Practical Handbook of Seawater Analysis, J.D.H. Strickland and T.R. Parsons, Bulletin 167, Fisheries Reasearch Board of Canada, Ottawa, Canada. 1968.

Chlorophyll <u>a</u> was determined by extraction of millipore filtered samples in 90% acetone and read spectrophotometrically.

11. Turbidity was determined with a Hach photoelectric nephelometer, calibrated in Jackson Turbidity Units.

12. Dissolved Oxygen

Reference: Standard Methods for the Examination of Water and Wastewater, 12 ed., 1965.

Dissolved Oxygen was determined by the azide modification of the basic Winkler method with the titration done potentiometrically with a Fisher automatic "titralyzer."

II STATION LOCATIONS

Station	Location
E 1	Route 50 Bridge
E 2	Queen Anne's Bridge
E 3	Bell's Junk Yard
E 4	Trailer Court
E 5	Wayson's Corner, Route 4 Bridge
E 5A	Western Branch
E 6	Mouth of Western Branch
E 6A	Middle of Jug Bay
E 7	Mouth of Lyon's Creek
E 8	Nottingham
E 9	Lower Marlboro, opposite wharf
E 10	High Power Lines
E 11	Halfway between Trueman Point and Deep Landing
E 12	500 yards east of PEPCO Canal
E 13	Chalk Point
E 14	Benedict Bridge Channel
E 15	Sheridan Point, Buoy 21

III SURVEY RESULTS 1969 PATUXENT ESTUARY ANNAPOLIS FIELD OFFICE

Sample Number	Date Sample Taken	Time Sample Taken	Water Temp °C	Secchi Disk Inches	Conduc- tivity µmhos	Salinity °/oo	Gage Height Feet	Total P PO _{li} mg/l	Inorganic P PO _l mg/l	TKN mg/l N	NO ₂ +NO ₃ mg/1 NO ₃ -N	NH3 mg/1 N	Chloro- phyll <u>a</u> µg/l	Tur- b idity JTU	DO mg/l
69020301 060901 070201 082501 102701 121801	2-03 6-09 7-02 8-25 10-27 12-18	0945 0945 0915 1045 0925 0910	4.5 20 22.9 20 10			Station E 1 1.1 0	- U.S. F 4.50 3.70 3.39	2.286 3.632 5.206 2.530 4.85 3.08	1.735	1.883 2.843 1.744 1.413	1.224 1.905 3.496 1.75 1.75	1.160 1.033 .780 .77 1.95 2.27	2.25 6.75 11.25 6.00 ND 7.50	17.5 72	10.37 4.15 2.62 5.75 7.07 11.32
69020302 060902 070202 082502 102702 121802	2-03 6-09 7-02 8-25 10-27 12-18	1010 1145 0850 1025 1055 1035	5 21 23 19.5 9.5			Station E 2	- Queen	Anne's Br 1.429 3.027 3.619 2.130 4.74 2.70	3.260 >3.0 2.28	1.463 1.529 .943 1.250	.997 2.543 3.248 1.64 > 1.80 1.04	.835 .948 .360 .73 1.74 1.45	1.50 4.50 9.75 3.75 ND 3.75	29 25	10.51 4.50 4.22 6.13 LA 11.58
69020303 060903 070203 082503 102703 121803	2-03 6-09 7-02 8-25 10-27 12-18	1040 1120 0830 1010 1115 1050	4.5 20.5 27 20 9			Station E 3	- Bell's	Junk Yard 1.771 2.368 2.984 2.032 4.31 2.42	- 2.633 >3.0 2.00	1.790 1.674 .551 1.069	.906 2.270 2.744 1.65 > 1.80 .969	.923 .585 .116 .64 1.38 1.46	5.25 7.50 11.25 3.00 MD 6.00	36 24	10.27 4.88 4.99 6.09 7.94 11.61

LA Laboratory accident

ND Not detectable

Sample Number	Date Sample Taken	Time Sample Taken	Water Temp °C	Secchi Disk Inches	Conduc- tivity umhos	Salinity °/oo	Gage Height Feet	Total P PO _L mg/l	Inorganic P PO, mg/1	TKN mg/l N	NO ₂ +NO ₃ mg/l NO ₃ -N	NH3 mg/l N	Chloro- phyll <u>a</u> µg/l	Tur- bidity JTU	DO mg/l
69020304 060904 070204 082504 102704 121804	2-03 6-09 7-02 8-25 10-27 12-18	1055 1110 0800 1000 1135 1110	4.5 21 27 20.5 11			Station E 4	- Traile	.892 2.638 1.725 1.395 2.82 2.09	1.299 2.12 1.64	1.265 1.773 1.273 1.456	1.302 2.230 2.212 1.22 1.75 1.01	.648 .693 .252 .39 .63	9.75 9.75 7.50 3.75 ND 3.75	14 27	8.85 4.42 3.87 5.50 8.41 11.32
69020305 060905 070205 082505 102705 121805	2-03 6-09 7-02 8-25 10-27 12-18	1140 1045 0730 0945 1015	5 21.5 26 21 9.5 1	14		Station E 5	- Route	4 Bridge, 1.543 1.751 1.196 1.514 3.73 1.92	.866 2.73 1.26	1.222 1.291 .858 1.238	1.052 1.954 2.720 1.26 1.66 1.18	1.025 .660 .230 .40 .87 1.96	3.00 5.25 5.25 8.25 3.75 3.75	25 21	9.61 3.97 3.20 5.21 7.94 11.03
69020306 060906 070206 082506 102706 121806	2-03 6-09 7-02 8-25 10-27 12-18	1255 1105 0745 1000 1035 1035	2.4 22 22 21 9.5 1.5	33 13		Station E 5.	A – Weste	290 -844 -693 -581 -858 1-53	.188 .400 1.11	1.049 .756 1.170 .400	.548 1.006 2.212 .20 .49	.145 .423 .030 .13 .17 .873	2.25 6.00 10.50 3.75 MD 3.00	12 19	10.61 3.61 6.33 4.48 8.70 11.24
69020307 060907 070207 082507 102707 121807	2-03 6-09 7-02 8-25 10-27 12-18	1245 1110 0750 1005 1045 1040	2.5 22.5 28 22 10 1	27 18		Station E 6	- Mouth	of Westerr .994 .688 .488 1.006 2.20 1.49	.125 1.33 .915	1.049 LA .790 .806	.937 .727 .006 .87 1.34 .885	.733 .343 .010 .35 .42 .820	4.50 14.25 16.50 8.25 .75 6.75	15 17.5	9.74 3.38 6.60 4.45 7.11 11.20
69020308 060908 070208 082508 102708 121808	2-03 6-09 7-02 8-25 10-27 12-18	1235 1121 0805 1020 1058 1050	2.4 24 28 23.5 11	25 20		Station E 6	A - Middl		.095 .408 .625	.926 .802 .886 .781	.898 .371 ND .46 .85	.833 .190 .005 .23 .28	2.25 44.25 41.25 6.75 24.25 5.25	14.5 27.5	9.34 5.34 6.79 3.93 10.09 11.55

LA Laboratory accident ND Not detectable

Sample Number	Date Sample Taken	Time Sample Taken	Water Temp °C	Secchi Disk Inches	Conduc- tivity umhos	Salinity °/oo	Gage Height Feet	Total P PO _l mg/l	Inorganic P PO _l mg/l	TKN mg/l N	NO ₂ +NO ₃ mg/1 NO ₃ -N	NH mg/1 N	Chloro- phyll <u>a</u> µg/l	Tur- bidity JTU	DO mg/1
69020309 060909 070209 082509 102709 121809	2-03 6-09 7-02 8-25 10-27 12-18	1215 1128 0810 1035 1105 1100	2.3 24 28 23.5 11	18 [°] 20		Station E 7	- Mouth	of Lyon's	.072 .152 .660	1.068 .640 .830 .756	.888 .340 ND .35 .39 .864	.663 .310 .008 .21 .29 .476	6.75 54.00 51.75 8.25 22.50 16.50	23 34	9.66 5.86 6.34 3.96 11.13 10.98
69020310 060910 070210 082510 102710 121810	2-03 6-09 7-02 8-25 10-27 12-18	1140 1145 0830 1050 1120 1115	2 24 28 25 12	12 18		13.9 2.5	- Nottin	gham815398438624539 1.28	.116 .145 .625	. 864 . 849 . 608 . 800 . 346	.781 .163 ND .36 .12 .802	.545 .218 .026 .23 .08 .476	6.75 41.25 61.50 15.75 9.00 30.00	35 18	10.36 6.64 5.66 4.63 9.65 10.89
69020311 060911 070211 082511 102711 121811	2-03 6-09 7-02 8-25 10-27 12-18	1125 1205 0850 1115 1140 1135	2 24.5 28 27 13 2	14 18	.6	Station E 9	- Lower	Marlboro, .405 .292 .447 .612 .309	opposite wharf	.883 .750 .699 .925	.698 .055 ND .40 .C3	.495 .173 ND .18 .05 .489	5.25 33.00 52.50 20.25 9.00 15.00	25 16.5	10.25 6.21 6.23 5.19 8.59 10.61
69020312 060912 070212 082512 102712 121812	2-03 6-09 7-02 8-25 10-27 12-18	1115 1220 0915 1135 1200 1155	2.2 25 28 27.5 13.5	14 22	2.9	Station E 10 1.0 8.6 6.7	O - High	Power Line .226 .282 .370 .560 .251	.149 .117 .223	.654 .558 .727 1.200	.581 .018 ND .33 .03 .724	.533 .158 MD .21 .06 .474	3.75 29.25 133.50 13.50 .75 LA	17 12	10.09 6.76 7.37 5.47 8.46 10.85
69020313 060913 070213 082513 102713 121813	2-03 6-09 7-02 8-25 10-27 12-18	1110 1230 0925 1145 1210 1210	2.6 25.5 29 28 14 2	19 30	4.1	Station E 1: 2.0 9.8 8.8	l <u>- Halîw</u>	211 .248 .371 .304 .279 .277	.166 .078 .150	. and Deep .512 .413 .574 .650	Landing .611 ND ND .21 .04 .452	.053 ND .26 .04 .305	ND 27.75 47.75 15.00 3.75 11.25	12.5 12	10.00 6.76 6.89 6.14 8.57 10.93

LA Laboratory accident ND Not detectable

Sample Number	Date Sample Taken	Time Sample Taken	Water Temp *C	Secchi Disk Inches	Conduc- tivity µmhos	Salinity */oo	Gage Height Feet	Total P PO _l mg/l	Inorganic P PO _{li} mg/l	TKN mg/1 N	NO ₂ +NO ₃ mg/1 NO ₃ -N	NH ₃	Chloro- phyll a µg/l	Tur- bidity JTU	DO mg/1
69020314 060914 070214 082514 102714 121814	2-03 6-09 7-02 8-25 10-27 12-18	1105 1245 0930 1155 1230 1215	3.5 25.5 29 28.5 14 2	5Ħ 5Ħ	6.0	Station E 12 3.6 11.6 11.3	e – 500 y	.198 .258 .392 .405 .278	.171 .102 .128	.691 .401 .568 1.244	.153 ND ND .10 .005 .293	.135 .415 .005 .10 .03	ND 16.50 51.00 6.75 .75 19.50	13 8.4	9.69 6.76 6.17 7.60 8.76 11.19
69020315 060915 070215 082515 102715 121815	2-03 6-09 7-02 8-25 10-27 12-18	1055 1250 0940 1205 1240 1225	3.7 29 28.5 14 2	28 42	9.4	Station E 13 6.9 12.5 11.7	3 - Chalk	201 .201 .224 .415 .545 .278 .221	.162 .127 .121	.593 .413 .614 .800	.375 ND ND .02 .005 .336	.118 .278 .008 .04 .04	4.50 15.75 32.25 12.00 .75 2.25	10 6.4	9.69 6.63 6.38 9.51 8.91 11.25
69020316 060916 070216 082516 102716 121816	2-03 6-09 7-02 8-25 10-27 12-18	1045 1305 0950 1220 1250 1235	3.5 24 28.5 28 14 2	36 46	10.6	Station E 14 8.1 12.8 12.1	: - Be ned	.139 .229 .332 .389 .251	.150 .137 .068	. 186 . 149 . 456	.382 ND ND .02 .005	.073 .343 .005 .14 .03	9.00 17.25 30.00 12.00 3.75 22.50	8 7	9.85 6.83 6.21 7.06 8.99 11.55
69020317 060917 070217 082517 102717	2-03 6-09 7-02 8-25 10-27	1030 1320 1015 1230 1305	2.1 28 28 14	75 46	. 11.5	Station E 15 9.3 13.8 13.7	- Sheri	dan Point, .121 .187 .540 .417 .241	.231 .127	.716 .401 1.131 .500	.265 ND ND .03 ND	.043 .423 .063 .05	8.25 14.25 63.75 21.00 3.75	4.5 3.2	10.13 6.44 8.74 7.58 9.14

ND Not detectable

